

International Workshop on Computational Nanotechnology

P:29 Hole Trapping in Amorphous HfO_2 and Al_2O_3 as a source of positive charging

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Positive charging in Si/SiO₂/HfO₂ stacks is usually associated with defects in the interface SiO₂ layer. Implementation of high mobility substrates, such as Ge and III-Vs, will see the gate dielectric exposed to hole trapping. Thin HfO₂ and Al₂O₃ layers deposited on, for example, Ge are usually amorphous [1]. To understand the ultimate behaviour of high- κ dielectrics in such devices we compare hole trapping in crystalline and amorphous HfO₂ and Al₂O₃ using DFT modelling. Nine periodic models of a-HfO₂, each containing 324 atoms, have been generated using classical MD and a melt and quench procedure. Further optimization of the volume and geometry of these structures was performed using DFT as implemented in the CP2K code [2] with the PBE0-TC-LRC functional. The model a-Al₂O₃ structures produced using the same method and contained 360 atoms.

The trapping of hole polarons on O atoms in m-HfO₂ and α -Al₂O₃ has been previously predicted theoretically [3, 4]. Recent calculations [5] predicted that holes can trap only at three-coordinated O sites in the bulk of m-HfO₂ with a trapping energy of 0.18 eV and with much larger trapping energies at surfaces, featuring two-coordinated O sites. Structural variations, such as under-coordination or elongated bonds, serve as precursor sites where holes can trap spontaneously in a-HfO₂ and a-Al₂O₃ [6]. In both materials over 90% of the hole spin density is localized on two O atoms. The characteristic atomic configurations of the polaron are shown in Fig. 1. The average hole trapping energies (thermal ionization energies) are 1.4 eV in a-HfO₂ and 1.2 eV in a-Al₂O₃. The distribution of hole trapping energies is shown in Fig. 3. We also find that both cells contain up to four precursor sites.

In m-HfO₂ two hole polarons form a stable bi-polaron state with the binding energy of 0.65 eV. This state is stabilized by forming of an O-O dimer. In a-HfO₂ the formation of hole bi-polaron takes place spontaneously at structural precursor sites. In both the amorphous and monoclinic cases, the O-O dimer formed has a bond length of 1.5 Å. Hole bi-polarons also form in a-Al₂O₃ again with an O-O bond length of 1.5 Å. The relaxation of surrounding ions is asymmetric (Fig. 2). In both oxides this involves ionic displacements exceeding 0.5 Å. This large structural reorganisation means that Hf-O, or Al-O, bonds are greatly weakened, suggesting the possibility of further defect creation.

2. FIGURES

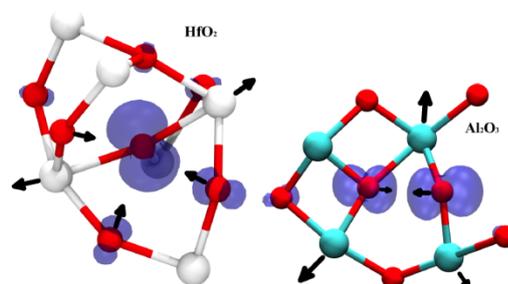
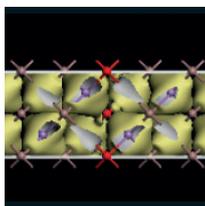


FIGURE 1. Configurations of hole polarons in a-HfO₂ and a-Al₂O₃. Red indicates O ions, cyan Al ions and white Hf ions. Spin density is indicated by the blue iso-surfaces. Arrows indicate direction of ionic displacements due to polaron trapping.



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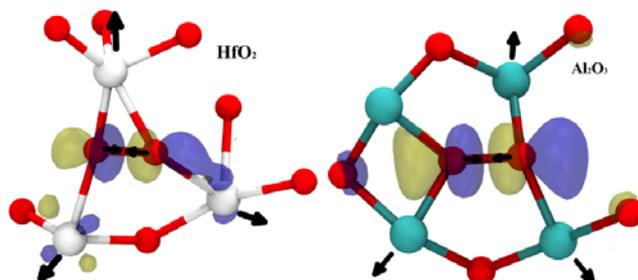


FIGURE 2. Configurations of bi-polarons in $\alpha\text{-Al}_2\text{O}_3$ and $\alpha\text{-HfO}_2$. The colored surfaces correspond to the unoccupied KS state of the bipolaron, with blue indicating a positive isovalue and yellow indicating a negative isovalue.

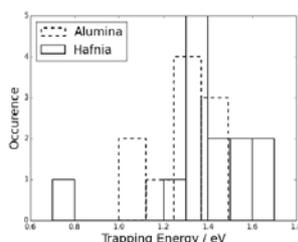


FIGURE 3. Trapping energies of single hole polarons in $\alpha\text{-Al}_2\text{O}_3$ and HfO_2 . Energies given are referenced to delocalized hole at the top of the valence band.

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